We claim:

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- 1 A process for the surface modification of a polymer substrate, the process comprising the steps of:
- a. absorbing a swelling monomer into the polymer substrate in order to swell the polymer substrate;
 - b. polymerizing the swelling monomer for a period of time;
- c. removing the polymer substrate from the swelling monomer at the end of the period of time.
- 2. The process of claim 1, wherein the polymer substrate is selected from the group consisting of: acrylics, acrylonitrile-butadiene-styrene copolymer, chlorinated polyvinylchloride, EPDM rubber, natural rubber, neoprene, nitrile rubber, polyethylene, polypropylene, polystyrene, polyurethanes, polyvinylchloride, silicones, thermoplastic elastomers, and vinylidene fluoride-hexafluoropropylene copolymer.
- 3. The process of claim 1, wherein the swelling monomer is absorbed into the polymer substrate in the presence of a solvent.
- 4. The process of claim 1, wherein the swelling monomer comprises at least one crosslinking monomer.
- 5. The process of claim 1, wherein the swelling monomer comprises at least one functional monomer.
- 6. The process of claim 1, wherein the swelling monomer comprises at least one crosslinking monomer and at least one functional monomer.
- 7. The process of claim 1, wherein the swelling monomer is selected from the group consisting of: acrylamides, methacrylamides, allyl crosslinkers, acrylates, methacrylates, and vinyl crosslinkers.
- 8. The process of claim 1, wherein the absorption of the swelling monomer occurs for a period of time that is less than or equal to approximately 96 hours.

- 9. The process of claim 1, wherein the absorption of the swelling monomer occurs at a temperature of between approximately -20°C to 150°C.
- 10. The process of claim 1, wherein an initiator initiates the polymerization of the swelling monomer.
 - 11. The process of claim 10, wherein the initiator is ultraviolet radiation.
 - 12. The process of claim 10, wherein the initiator is heat.
 - 13. The process of claim 10, wherein the initiator is ionization radiation.
 - 14. The process of claim 10, wherein the initiator is a chemical catalyst.
- 15. The process of claim 14, wherein the initiator is selected from the group consisting of: azo-initiators, peroxide initiators, and UV/visible initiators.
- 16. The process of claim 14, wherein the initiator is present in the aqueous solution in a concentration of less than or equal to 10%.
- 17. The process of claim 1, wherein the polymerization step is carried out for a period of time between approximately 10 seconds to 72 hours.
- 18. The process of claim 1, wherein the polymerization step is carried out at a temperature of between approximately -78°C to 150°C.
 - 19. The process of claim 1, wherein the polymer is silicone.
- 20. The process of claim 19, wherein heparin will adhere to the surface of the silicone polymer following its removal from the swelling monomer.
- 21. A process for the surface modification of a polymer substrate, the process comprising the steps of:

- a. absorbing a swelling monomer into the polymer substrate for a period of time in order to swell the polymer substrate;
 - b. removing the swollen polymer from the swelling monomer;
 - c. transferring the swollen polymer to a reaction mixture

containing at least one functional monomer;

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- d. polymerizing the functional monomer in the reaction mixture containing the swollen polymer substrate for a period of time; and
 - e. removing the polymer from the reaction mixture.
- The process of claim 21, wherein the polymer substrate is selected from the group consisting of: acrylics, acrylonitrile-butadiene-styrene copolymer, chlorinated polyvinylchloride, EPDM rubber, natural rubber, neoprene, nitrile rubber, polyethylene, polypropylene, polystyrene, polyurethanes, polyvinylchloride, silicones, thermoplastic elastomers, and vinylidene fluoride-hexafluoropropylene copolymer.
- 23. The process of claim 21, wherein the swelling monomer is absorbed into the polymer substrate in the presence of a solvent.
- 24. The process of claim 21, wherein the swelling monomer comprises at least one crosslinking monomer.
- 25. The process of claim 21, wherein the swelling monomer comprises at least one functional monomer.
- 26. The process of claim 21, wherein the swelling monomer comprises at least one crosslinking monomer and at least one functional monomer.
- 27. The process of claim 21, wherein the swelling monomer is selected from the group consisting of: acrylamides, methacrylamides, allyl crosslinkers, acrylates, methacrylates, and vinyl crosslinkers.
- 28. The process of claim 21, wherein the absorption of the swelling monomer occurs for a period of time that is less than or equal to approximately 96 hours.
- 29. The process of claim 21, wherein the absorption of the swelling monomer occurs at a temperature of between approximately -20°C to 150°C.
- 30. The process of claim 21, wherein an initiator initiates the polymerization of the swelling monomer.

- 31. The process of claim 30, wherein the initiator is ultraviolet radiation.
- 32. The process of claim 30, wherein the initiator is heat.
- 33. The process of claim 30, wherein the initiator is ionization radiation.
- 34. The process of claim 30, wherein the initiator is a chemical catalyst.
- 35. The process of claim 34, wherein the initiator is selected from the group consisting of: azo-initiators, peroxide initiators, and UV/visible initiators.
- 36. The process of claim 34, wherein the initiator is present in the aqueous solution in a concentration of less than or equal to 10%.
- 37. The process of claim 21, wherein the polymerization step is carried out for a period of time between approximately 10 seconds to 72 hours.
- 38. The process of claim 21, wherein the polymerization step is carried out at a temperature of between approximately -78°C to 150°C.
 - 39. The process of claim 21, wherein the polymer is silicone.
- 40. The process of claim 39, wherein heparin will adhere to the surface of the silicone polymer following its removal from the swelling monomer.
- 41. The process of claim 21, wherein the functional monomer has at least one amine group.
- 42. The process of claim 21, wherein the functional monomer has at least one hydroxyl group.
- 43. The process of claim 21, wherein the functional monomer has at least one carboxyl group.
- 44. The process of claim 21, wherein the functional monomers are selected from the group consisting of: acrylamides, methacrylamides, acrylates, methacrylates, allyl monomers, vinyl monomers, and styrenic monomers

- The process of claim 21, wherein the functional monomers are present in the reaction mixture in a concentration of between approximately 0.01% to 100%.
- 46. The process of claim 21, wherein the reaction mixture further contains at least one solvent.
- 47. A process for the surface modification of a silicone substrate, the process comprising the steps of:
- a. absorbing ethylene glycol dimethacrylate into silicone for between approximately 0.1 hours to 72 hours at a temperature of between approximately 0°C and 100°C in order to swell the silicone;
- b. removing the swollen silicone from the ethylene glycol dimethacrylate;

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- c. transferring the swollen silicone into an aqueous solution containing 2-aminoethyl methacrylate hydrochloride in a concentration of between approximately 0.1% and 50% and 2,2'-azobis(2-methylpropionamidine) dihydrochloride in a concentration of between approximately 0.1% and 10%;
- d. contacting the swollen silicone with the 2-aminoethyl methacrylate hydrochloride and the 2,2'-azobis(2-methylpropionamidine) dihydrochloride at a temperature of between approximately 30°C and 80°C for between approximately 0.1 hours and 24 hours; and
 - e. removing the silicone from the aqueous solution.
- 48. A process for forming a surface interpenetrating polymer network on a silicone substrate, the process comprising the steps of:
- a absorbing bis(2-methacryloxyethyl) phosphate into silicone for between approximately 0.1 hours and 72 hours at room temperature in order to swell the silicone;
- b. removing the swollen silicone from the bis(2-methacryloxyethyl) phosphate;
- c. transferring the swollen silicone into an aqueous solution containing 2-aminoethyl methacrylate hydrochloride in a concentration of between approximately 0.1% and 50% and 2-hydroxy-2-methyl-1-phenylpropanone in a

concentration of between approximately 0.1% and 10%;

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- d. contacting the swollen polymer with the 2-aminoethyl methacrylate hydrochloride and the 2-hydroxy-2-methyl-1-phenylpropanone at a temperature of between approximately 30°C and 80°C for between approximately 1 minute to 10 hours with UV radiation; and
 - e. removing the silicone from the aqueous solution.
- 49. A process for the heparinization of a polymer treated with the process of claim 1, the process comprising the steps of:
- a. adding an amount of heparin sodium and citric acid to deionized water to create a heparin solution;
- b. adjusting the pH of the heparin solution to between approximately 2 and 5;
 - c. adding NaCNBH₃ to the heparin solution;
- d. adding a polymer substrate material that has previously undergone the process of claim 1 to the heparin solution;
 - e. reacting the polymer substrate material in the heparin solution;
 - f. removing the polymer material from the heparin solution;
- g. rinsing the polymer material with deionized water and a sodium borate solution;
- h. adding the polymer material into a sodium borate solution with polyethylenimine;
- i. removing the polymer material from the sodium borate solution and rinsing the polymer material with deionized water;
 - j. adding the polymer material back into the heparin solution;
 - k. reacting the polymer material in the heparin solution;
 - 1. removing the polymer material from the heparin solution; and
- m. rinsing the polymer material with a sodium borate solution and deionized water.
 - 50. The heparinized material of claim 49.
- 51. The process of claim 49, wherein the polymer is treated with the process of claim 21.

- 52. The heparinized material of claim 51.
- 53. The process of claim 49, wherein the polymer is treated with the process of claim 47.
 - 54. The heparinized material of claim 53.
- 55. The process of claim 49, wherein the polymer is treated with the process of claim 48.
 - 56. The heparinized material of claim 55.
 - 57. The process of claim 49, wherein the polymer substrate is silicone.
 - The heparinized silicone material formed by the process of claim 57.
- 59. The process of claim 49, wherein the polymer substrate is a silicone intraocular lens.
- 60. The heparinized silicone intraocular lens formed by the process of claim 59.
- The process of claim 49, wherein the polymer substrate is a silicone contact lens.
- 62. The heparinized silicone contact lens formed by the process of claim 61.
- 63. The polymer substrate having the surface modification formed by the process of claim 1.
- 64. The polymer substrate having the surface modification formed by the process of claim 21.
- The polymer substrate having the surface modification formed by the process of claim 47.
 - 66. The polymer substrate having the surface modification formed by the

process of claim 48.

- 67. A silicone material having the surface modification produced by the process of claim 1.
- 68. A silicone material having the surface modification produced by the process of claim 21.
- 69. A silicone material having the surface modification produced by the process of claim 47.
- 70. A silicone material having the surface modification produced by the process of claim 48.
- 71. A silicone intraocular lens having the surface modification formed by the process of claim 1.
- 72. A silicone intraocular lens having the surface modification formed by the process of claim 21.
- 73. A silicone intraocular lens having the surface modification formed by the process of claim 47.
- 74. A silicone intraocular lens having the surface modification formed by the process of claim 48.
- 75. A silicone contact lens having the surface modification formed by the process of claim 1.
- 76. A silicone contact lens having the surface modification formed by the process of claim 21.
- 77. A silicone contact lens having the surface modification formed by the process of claim 47.
- 78. A silicone contact lens having the surface modification formed by the process of claim 48.